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## Triterpenoids from *Uncaria rhynchophylla* and Their PTP1B Inhibitory Activity

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### ABSTRACT

*Uncaria rhynchophylla* (Gouteng) is a famous traditional Chinese medicine used for psychiatric and hypotensive purposes in China. In this study, the ethyl acetate (EtOAc) part of *U. rhynchophylla* was revealed with protein tyrosine phosphatase 1B (PTP1B) inhibitory activity. Subsequent investigation on the EtOAc part yielded one new triterpenoid, 3 $\beta$ -formyloxy-6 $\beta$ ,19 $\alpha$ -dihydroxyurs-12-en-28-oic acid (**1**) and four known ones, 3 $\beta$ ,6 $\beta$ ,19 $\alpha$ -trihydroxyurs-12-en-28-oic acid (**2**), 2-oxopomolic acid (**3**), 3 $\beta$ ,19 $\alpha$ -dihydroxy-6-oxo-olean-12-en-28-oic acid (**4**) and sumaresinolic acid (**5**). The structure of compound **1** was determined by extensive HRESIMS, IR, 1D and 2D NMR spectroscopic analyses. Two ursane-type triterpenoids (**2** and **3**) showed selective inhibition on PTP1B with IC<sub>50</sub> values of 48.2 and 178.7  $\mu$ M. The enzyme kinetic study suggested that compounds **2** and **3** were mix-type inhibitors on PTP1B with K<sub>i</sub> values of 15.6 and 132.5  $\mu$ M. This investigation manifests the antidiabetic potency of *U. rhynchophylla* with triterpenoids as the active constituents.

### KEYWORDS

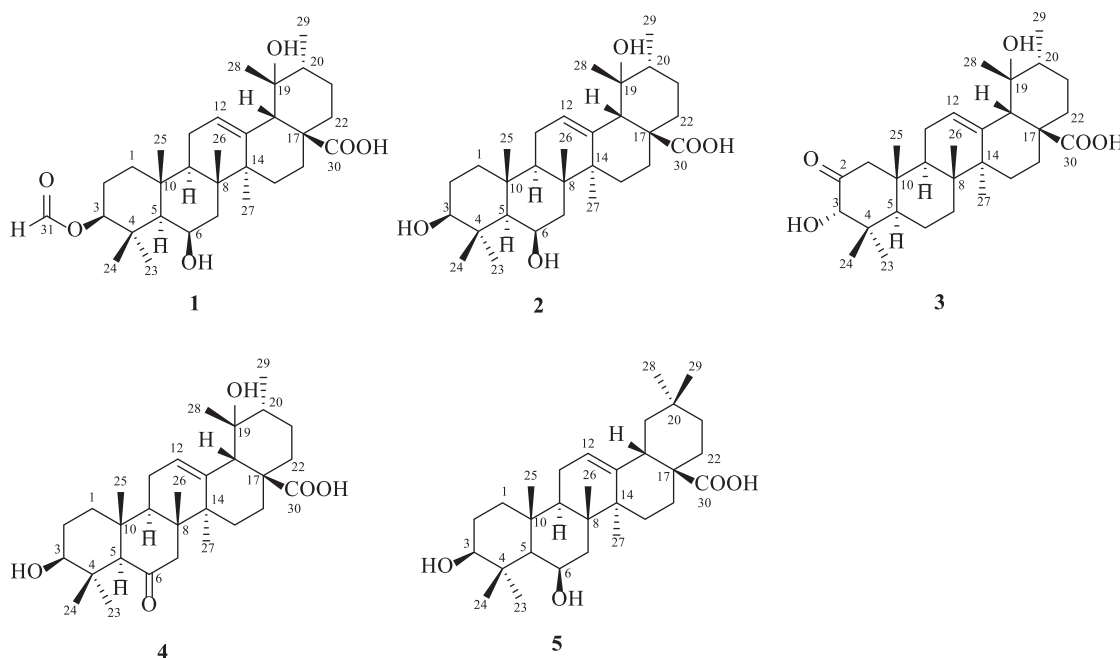
*Uncaria rhynchophylla*; triterpenoids; antidiabetic effects; PTP1B inhibitors

## 1 Introduction

Diabetes mellitus as one of the most serious chronic diseases in the world is characterized by high levels of blood glucose (hyperglycemia) [1]. Of the 425 million people suffering from diabetes mellitus, more than 90% cases are type 2 diabetes and lead to 1.6 million deaths every year. According to the World Health Organization (WHO), diabetes mellitus will reach to the seventh cause of death by 2030 [2]. Although many antidiabetic drugs with diverse mechanism of action are available on the market, the inevitable side effects including hypoglycemia, allergic reactions, weight gain, gastrointestinal reactions, liver damage and nutritional disorders hinder their application. Thus, different therapeutic targets are being gradually discovered along with the in-depth study of diabetes mellitus. Protein tyrosine phosphatase 1B (PTP1B), a negative regulator of insulin and leptin signal transduction, has been regarded as a hot target in treating diabetes mellitus. Recently, many kinds of PTP1B inhibitors including alkaloids, terpenoids, phenolics, fatty acids, and steroids have been reported, which provide valuable clues for the discovery of novel antidiabetic drugs [3,4]. However, it is a great challenge to discover inhibitors specific to each PTPs due to their high homology. Especially, T-cell protein tyrosine phosphatase (TCPTP) sharing about 74% identity with PTP1B in catalytic domain is a major obstacle in the search of PTP1B selective inhibitors.



Traditional Chinese medicines are important sources for drug discovery [5]. *Uncaria rhynchophylla* (Gouteng) is a famous traditional Chinese medicine used for psychiatric and hypotensive purposes in China [6,7]. Alkaloids, triterpenoids, flavonoids and organic acids are the main constituents in *U. rhynchophylla* [8]. Our previous study disclosed that different parts of *U. rhynchophylla* could be well differentiated by their chemical profiles [9], from which two pairs of dimeric isoechinulin-type alkaloids [10], two flavanols [11], and one indole alkaloid triglycoside [12] with antidepressant potency were isolated under the guidance of bioassay. As a comparison, the antidiabetic constituents of *U. rhynchophylla* have been little investigated. A recent study reported that hirsutine, the characteristic constituents in *Uncaria* plants, showed antidiabetic effects by regulating glucose homeostasis and ameliorating hepatic and cardiac insulin resistance in diabetic mice [13]. As a continuous search for PTP1B selective inhibitors from natural sources [14], this investigation yielded one new triterpenoid, 3 $\beta$ -formyloxy-6 $\beta$ ,19 $\alpha$ -dihydroxyurs-12-en-28-oic acid (1), along with four known ones, 3 $\beta$ ,6 $\beta$ ,19 $\alpha$ -trihydroxyurs-12-en-28-oic acid (2), 2-oxopomolic acid (3), 3 $\beta$ ,19 $\alpha$ -dihydroxy-6-oxo-olean-12-en-28-oic acid (4) and sumaresinolic acid (5) from *U. rhynchophylla* (Fig. 1). Herein, we describe their isolation, structural characterization, and inhibitory activities on PTP1B and TCPTP.



**Figure 1:** Structures of compounds 1–5

## 2 Materials and Methods

### 2.1 General Experimental Procedure

1D and 2D NMR spectra were performed on an Advance III-600 spectrometer (Bruker, Bremerhaven, Germany). HRESIMS data were recorded on a LCMS-IT-TOF mass spectrometer (Shimadzu, Kyoto, Japan). Thin-layer chromatography (TLC) analyses were performed on silica gel GF254 plates (Yantai Jiangyou Silicon Development Company, Yantai, China), and spots were detected by heating after spraying with 10% H<sub>2</sub>SO<sub>4</sub> in EtOH. Chromatographic silica gel (200~300 mesh, Qingdao Makall Chemical Company, Qingdao, China) and Sephadex LH-20 gel (GE Healthcare Bio-Sciences AB, Uppsala, Sweden) were used for chromatography.

## 2.2 Plant Materials

The aboveground parts of *Uncaria rhynchophylla* (Miq.) Miq. ex Havil. were collected from Pingxiang City, Jiangxi Province of China, which were authenticated by Dr. Li-Gong Lei (Kunming Institute of Botany, Chinese Academy of Sciences). A voucher specimen (2015080102) was deposited in the Laboratory of Antivirus and Natural Medicinal Chemistry, Kunming Institute of Botany, Chinese Academy of Sciences.

## 2.3 Extraction and Isolation

The air-dried sample (1.5 kg) was powdered and extracted with 70% aqueous EtOH (10 L) under reflux for two times (each 2 h). The combined EtOH extract was evaporated under reduced pressure and partitioned between H<sub>2</sub>O and EtOAc. The EtOAc part (48 g) was fractionated by silica gel column chromatography (Si CC) using a petroleum ether-acetone gradient (90:10, 80:20, 70:30, 90:40, 0:100, v/v) to give eight fractions, Frs. 1~8. Fr. 3 (14.7 g) was separated by Si CC and eluted with CHCl<sub>3</sub>-acetone to provide five subfractions, Frs. 3-1~3-5. Fr. 3-3 (3.8 g) was further chromatographed on a silica gel column using petroleum ether-EtOAc (70:30) to afford compounds **1** (0.7 g) and **2** (1.1 g). Fr. 3-4 (3.6 g) was purified by repeated Si CC using petroleum ether-EtOAc and CHCl<sub>3</sub>-acetone elution, and Sephadex LH-20 CC (CHCl<sub>3</sub>-MeOH) to afford compounds **3** (15.3 mg), **4** (21.5 mg), and **5** (20.0 mg).

**Compound 1**: white powder; IR (KBr)  $\nu_{\max}$ : 3632, 3550-3400 (broad peak), 1719, 1691, 1450, 1369, 1209, 1183, 1149, 1037, 968, 934, 903, 768, 656 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; HRESIMS  $m/z$  515.3382 ([M-H]<sup>-</sup>, C<sub>31</sub>H<sub>47</sub>O<sub>6</sub>, +0.4 mDa).

**Table 1:** <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (150 MHz) data for compound **1** in methanol-*d*<sub>4</sub>

No.	$\delta_{\text{H}}$	$\delta_{\text{C}}$	No.	$\delta_{\text{H}}$	$\delta_{\text{C}}$
<b>1</b>	1.38 (m) 2.06 (dd, 10.1, 4.6)	40.7 CH <sub>2</sub>	<b>17</b>	–	47.4 C
<b>2</b>	2.07 (m) 2.15 (m)	27.1 CH <sub>2</sub>	<b>18</b>	2.50 (s)	53.2 CH
<b>3</b>	4.58 (dd, 11.8, 4.4)	81.5 CH	<b>19</b>	–	73.1 C
<b>4</b>	–	38.5 C	<b>20</b>	1.49 (m)	41.2 CH
<b>5</b>	1.06 (m)	55.7 CH	<b>21</b>	1.33 (m) 2.10 (m)	25.5 CH <sub>2</sub>
<b>6</b>	4.57 (m)	68.2 CH	<b>22</b>	2.07 (m) 2.15 (m)	37.5 CH <sub>2</sub>
<b>7</b>	1.35 (m) 1.58 (m)	40.2 CH <sub>2</sub>	<b>23</b>	1.05 (s)	27.7 CH <sub>3</sub>
<b>8</b>	–	39.0 C	<b>24</b>	0.99 (s)	17.6 CH <sub>3</sub>
<b>9</b>	1.93 (dd, 7.4, 3.2)	47.6 CH	<b>25</b>	1.02 (s)	16.1 CH <sub>3</sub>
<b>10</b>	–	36.4 C	<b>26</b>	1.09 (s)	18.1 CH <sub>3</sub>
<b>11</b>	1.87 (m) 2.08 (m)	23.6 CH <sub>2</sub>	<b>27</b>	1.71 (s)	23.9 CH <sub>3</sub>
<b>12</b>	5.39 (t, 3.4)	129.1 CH	<b>28</b>	1.43 (s)	26.1 CH <sub>3</sub>
<b>13</b>	–	137.4 C	<b>29</b>	1.11 (d, 6.0)	16.8 CH <sub>3</sub>

(Continued)

Table 1 (continued)					
No.	$\delta_{\text{H}}$	$\delta_{\text{C}}$	No.	$\delta_{\text{H}}$	$\delta_{\text{C}}$
<b>14</b>	–	41.8 C	<b>30</b>	–	181.4 C
<b>15</b>	1.26 (m) 2.31 (brd, 4.6)	28.2 CH <sub>2</sub>	<b>31</b>	8.14 (s)	161.6 CH
<b>16</b>	2.10 (m) 2.50 (m)	24.4 CH <sub>2</sub>			

## 2.4 Bioassay in Vitro

Inhibitory activity against PTP1B and TCPTP was performed in accordance with our previous report [15]. In brief, a mixture containing tested samples, enzymes, and working buffer was transferred into 96-well plates and preincubated at 37°C. The reaction was initiated by adding *p*-nitrophenyl phosphate, and further incubated for 30 min. The reaction was stopped by adding 100  $\mu\text{L}$  of Na<sub>2</sub>CO<sub>3</sub> solution (0.1 M). The cell plates were monitored at 405 nm using a BIO-RAD Model 680 microplate reader (Bio-Rad Laboratories, CA, USA). Sodium orthovanadate was used as the positive control.

## 3 Results

### 3.1 Structural Identification

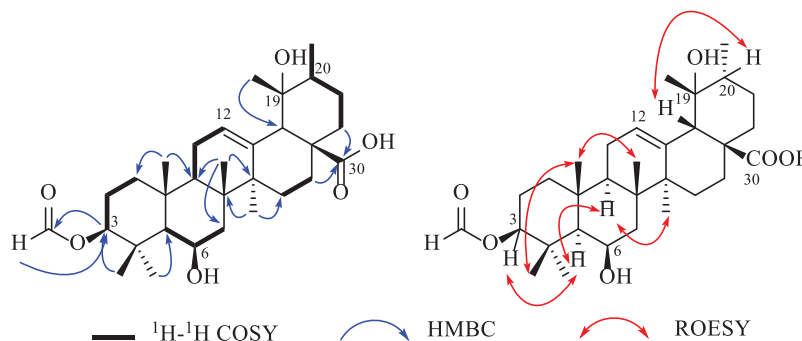
Compound **1** was deduced with the molecular formula of C<sub>31</sub>H<sub>48</sub>O<sub>6</sub> by the deprotonated ion at *m/z* 515.3382 ([M–H]<sup>–</sup>, +0.4 mDa) in the high-resolution electrospray ionization mass spectrometry (HRESIMS), suggesting eight indices of hydrogen deficiency. Its IR spectrum indicated the presence of hydroxyl (3632 cm<sup>–1</sup>), carbonyl (1719 cm<sup>–1</sup>) and formyl (1691 cm<sup>–1</sup>) groups. In accordance with the molecular formula, all the 31 carbons were well recognized in the <sup>13</sup>C NMR (DEPT) spectrum, including seven methyls, eight methylenes, eight methines, and eight non-protonated carbons. In the <sup>1</sup>H NMR spectrum, seven methyls at  $\delta_{\text{H}}$  1.71 (3H, s, H-27), 1.43 (3H, s, H-28), 1.11 (3H, *J* = 6.0 Hz, H-29), 1.09 (3H, s, H-26), 1.05 (3H, s, H-23), 1.02 (3H, s, H-25), and 0.99 (3H, s, H-24), one olefinic proton at  $\delta_{\text{H}}$  5.39 (t, *J* = 3.4 Hz), and two oxygenated methines at  $\delta_{\text{H}}$  4.58 (1H, dd, *J* = 11.8, 4.4 Hz) and 4.57 (1H, m) were obviously recognized. Taking the characteristic carbons of seven methyls at  $\delta_{\text{C}}$  27.7, 26.1, 23.9, 18.1, 17.6, 16.8, and 16.1, one tri-substituted double bond at  $\delta_{\text{C}}$  129.1 and 137.1, one carboxyl at  $\delta_{\text{C}}$  181.4 into consideration, an ursane-type triterpenoid was deduced. The <sup>1</sup>H and <sup>13</sup>C NMR data (Table 1) of compound **1** were similar with those of 3 $\beta$ ,6 $\beta$ ,19 $\alpha$ -trihydroxyurs-12-en-28-oic acid (**2**) [16], except for an extra formyl group ( $\delta_{\text{C}}$  161.6 and  $\delta_{\text{H}}$  8.14) in **1**, and the de-shielded shift of C-3 from  $\delta_{\text{C}}$  79.4 to 81.5. Thus, compound **1** was reasonably proposed to be the 3-O-formyl derivative of **2**. This deduction was fully confirmed by the HMBC correlations from H-31 to C-3 and from H-3 to C-31. The  $\alpha$ -orientation of H-3 was verified by the dd splitting (*J* = 11.8, 4.4 Hz) and the ROESY correlation between H-3 and H-5 (Fig. 2). Through the above analysis, the structure of **1** was determined to be 3 $\beta$ -formyloxy-6 $\beta$ ,19 $\alpha$ -dihydroxyurs-12-en-28-oic acid.

The known compounds were determined as 3 $\beta$ ,6 $\beta$ ,19 $\alpha$ -trihydroxyurs-12-en-28-oic acid (**2**) [16], 2-oxopomolic acid (**3**) [17], 3 $\beta$ ,19 $\alpha$ -dihydroxy-6-oxo-olean-12-en-28-oic acid (**4**) [18] and sumaresinolic acid (**5**) [19], by comparing with the spectroscopic data reported in literatures.

### 3.2 PTP1B Inhibitory Activity

In this study, the ethanol extraction of *U. rhynchophylla* was revealed with PTP1B inhibition with inhibitory ratios of 44.8% and 14.3% at concentrations of 400 and 200  $\mu\text{g/mL}$ , respectively. After

extracted with EtOAc, the EtOAc part showed increased activity with inhibitory ratios of 89.8% and 22.5%, whereas the water part maintained the similar activity with the ethanol extraction (Table 2).



**Figure 2:** Key 2D NMR correlations of compound **1**

**Table 2:** Inhibitory activities of *U. rhynchophylla* against PTP1B<sup>a</sup>

Parts	PTP1B inhibitory ratio (%)	
	400 $\mu\text{g/mL}$	200 $\mu\text{g/mL}$
Ethanol extraction	44.8 $\pm$ 4.8	14.3 $\pm$ 1.1
EtOAc part	89.8 $\pm$ 1.3	22.5 $\pm$ 4.2
Water part	50.2 $\pm$ 3.9	9.7 $\pm$ 2.1

Note: <sup>a</sup>Sodium orthovanadate was used as the positive control ( $\text{IC}_{50} = 191.0 \mu\text{M}$ ).

In order to evaluate the antidiabetic effects of the isolates, compounds **1–5** were assayed for their inhibition on PTP1B and TCPTP. As shown in Table 3, compound **2** showed significant inhibition on PTP1B with inhibitory ratios of 96.9% and 87.6% at concentrations of 200 and 100  $\mu\text{M}$ , respectively. Compounds **3** and **4** exhibited moderate activity against PTP1B with inhibitory ratios of 56.2% and 29.8% at 200  $\mu\text{M}$ , and 36.3% and 21.7% at 100  $\mu\text{M}$ , whereas compounds **1** and **5** were inactive at the tested concentrations. All the isolates showed moderate to weak inhibition on TCPTP, indicating a PTP1B selectivity.

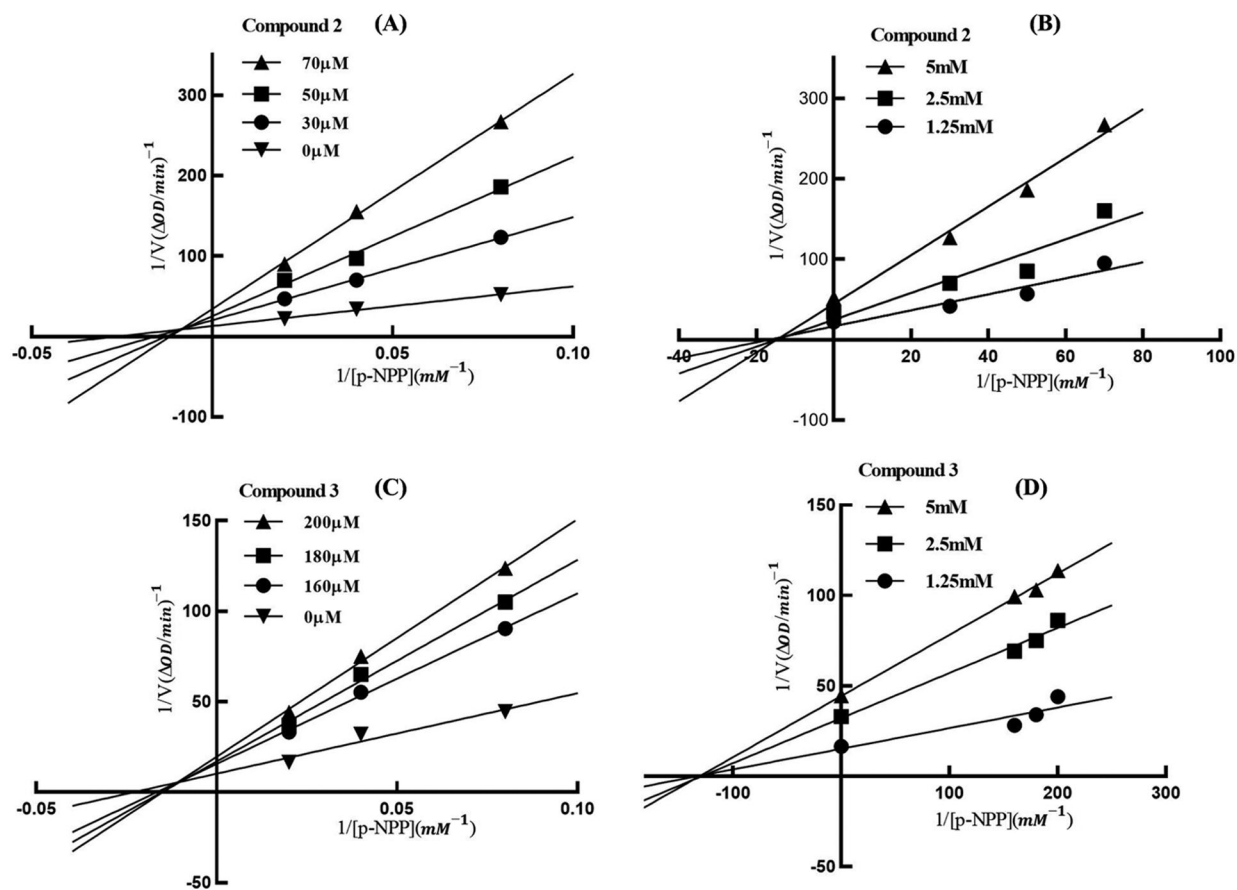
**Table 3:** Inhibitory activities of compounds **1–5** against PTP1B and TCPTP<sup>a</sup>

Compounds	PTP1B inhibitory ratio (%)		$\text{IC}_{50}$ ( $\mu\text{M}$ )	TCPTP inhibitory ratio (%)		$\text{IC}_{50}$ ( $\mu\text{M}$ )
	200 $\mu\text{M}$	100 $\mu\text{M}$		200 $\mu\text{M}$	100 $\mu\text{M}$	
<b>1</b>	6.6 $\pm$ 1.5	3.1 $\pm$ 6.6	/	21.6 $\pm$ 5.5	0.1 $\pm$ 3.0	/
<b>2</b>	96.9 $\pm$ 1.0	87.6 $\pm$ 2.0	48.2	32.4 $\pm$ 2.1	14.4 $\pm$ 4.2	/
<b>3</b>	56.2 $\pm$ 4.1	36.3 $\pm$ 3.5	178.7	28.7 $\pm$ 3.0	13.2 $\pm$ 4.8	/
<b>4</b>	29.8 $\pm$ 3.1	21.7 $\pm$ 4.7	/	33.3 $\pm$ 2.2	19.5 $\pm$ 1.7	/
<b>5</b>	1.7 $\pm$ 0.9	2.6 $\pm$ 3.8	/	9.8 $\pm$ 1.3	0.9 $\pm$ 3.1	/

Note: <sup>a</sup>Sodium orthovanadate was used as the positive controls,  $\text{IC}_{50} = 192.3$  (PTP1B) and 183.7 (TCPTP)  $\mu\text{M}$ .

The dose-effect relationships of compounds **2** and **3** on PTP1B were further evaluated at five concentrations to provide their respective  $IC_{50}$  value of 48.2 and 178.7  $\mu\text{M}$ , more potent than the positive control, sodium orthovanadate ( $IC_{50}$ , 192.3  $\mu\text{M}$ ).

The inhibition mode of compounds **2** and **3** on PTP1B was studied using Lineweaver-Burk double reciprocal and Dixon single reciprocal plots (Fig. 3). In the Lineweaver-Burk plots, the lines of compounds **2** and **3** intersected on the second quadrant, indicating the mixed-type inhibition. The  $K_i$  values of compounds **2** and **3** were calculated as 15.6 and 132.5  $\mu\text{M}$  by the Dixon plots.



**Figure 3:** Lineweaver-Burk (A/C) and Dixon (B/D) plots of compounds **2** and **3** against PTP1B. In the Lineweaver-Burk plots (A/C), the lines of compounds **2** and **3** intersected on the second quadrant meaning that the  $K_m$  was increased but  $V_{max}$  was decreased with the increased concentration of inhibitors, which indicated **2** and **3** as mixed-type inhibitors

#### 4 Conclusion

In this investigation, the ethanol extraction of *U. rhynchophylla* was revealed with PTP1B inhibitory effects. Under the guidance of bioassay, one new and four known triterpenoids were isolated from the EtOAc part. Two ursane-type triterpenoids **2** and **3** showed selective inhibition on PTP1B with  $IC_{50}$  values of 48.2 and 178.7  $\mu\text{M}$ , respectively. The enzyme kinetic study manifested that compounds **2** and **3** were mix-type inhibitors on PTP1B with  $K_i$  values of 15.6 and 132.5  $\mu\text{M}$ . This investigation provides valuable clues for understanding the antidiabetic effects of *U. rhynchophylla* and searching new PTP1B selective inhibitors from natural sources.

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**Conflicts of Interest:** The authors declare that they have no conflicts of interest to report regarding the present study.

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