



REVIEW

Potential Economic Value of Chitin and Its Derivatives as Major Biomaterials of Seafood Waste, with Particular Reference to Southeast Asia

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ABSTRACT

With a growing population, changes in consumerism behavior and trends in consumption in Indo-Pacific Asia, our seafood processing and consumption practices produce a large volume of waste products. There are several advantages in regulating and sustaining shellfish processing industries. The major advantage of waste management is that it leads to better conservation of natural resources in the long run. Shrimp shell waste contains useful biomaterials, which are still untapped due to inadequate waste disposal and solid waste management. Chitin, the major component of shell waste, can be extracted either chemically or biologically. The chemical extraction approaches, which use acids and alkali, could be an environmental burden. On the other hand, biological methods can be eco-friendly alternatives for shell waste management. In this review, recent trends in management of shellfish waste as sources of chitin, conversion of chitin into chitosan, economic aspects of waste treatment and application of chitosan will be discussed.

KEYWORDS

Seafood waste; waste management; chitin; chitosan; economic value; Southeast Asian fisheries

1 Introduction

Asia and the Pacific play a critical role in agricultural productivity, and this area is very crucial for fishing and aquaculture activities. In addition, aquaculture is particularly highly diversified and contributes to the economic well-being of Southeast Asia [1,2]. In the year 2018, aquaculture production in South Asia was 13 million tons [3]. The overall inland and marine captures were 49.2 million tons in the year 2016, which increased to 49.9 million tons in 2018 [4]. The Southeast Asia region covers a large expanse of marine water and does not have any high sea areas. The South China Sea has a total area of approximately 3.3 million km². Many Southeast Asia's developing countries depend more on coastal fish farming than on marine resources, particularly the Philippines and Vietnam, which are frequently exposed to typhoons [5]. Contiguous to the South China Sea are the Malacca Straits, the Java, Flores, Banda,



Ceram, Molucca, Celebes and Sulu seas, and the Philippine Sea [6]. This semi-closed sea and its contiguous waters provide primary sources of food and protein for people, as well as sources of income and livelihoods for many people by generating several billion dollars in gross domestic product (GDP) for the region [1,7].

It is anticipated that total fish consumption and demand will grow with rapid population growth, which is estimated to grow at 2.3% annually between the years 2010 and 2030. This is likely to have an impact on both capture and aquaculture production, resulting in voluminous amounts of seafood discards. The global fishing and aquaculture sector, which has been growing since the 1970s, produces 6 to 8 million tons of waste each year [8]. Crustacean production in Asia was 14,111 thousand tons, while mollusk production was 31,671 tons [1,9].

How can the fishery waste be beneficially used? Till to date, some companies are producing pet food, fishmeal or fertilizer from seafood waste [10], even now a lot of seafood waste is being discarded as landfill. Although the fishing and aquaculture sector can be a tight margin sector, this would boost profitability through the creative recycling strategy which can be achieved to minimize waste. The objective of this article is to discuss the sustainable use of seafood shell resources, the recovery process of valuable ingredients, particularly chitin, and improve current practice or policy in waste management. The study also discusses the relationships and applications of shell waste, including chitin, calcium, protein, and carotenoids. In addition, current insights on the preparation of chitin nanomaterials are discussed along with their practical applications. Subsequently, the promising applications of chitin-based nanotechnology, economic value, and the use of chitosan in the area are also pointed out.

2 Southeast Asia: Potential Seafood Waste from Seafood Consumption

Fishery activity can be viewed as a multi-gear and multi-species related process. Both inland and marine water fishery operations are carried out in a traditional way and carried out by small scale fishermen and family scale ventures. Only a small proportion of Southeast Asia's fishery industry has transitioned from small-scale capture fisheries processing, mostly marketed domestically, to larger-scale export-oriented fisheries in Asia, particularly in Southeast Asia, where despite being the largest producers of seafood goods, fishing families in many parts of rural areas remain socially and economically impoverished. It is only with technical assistance, increased investment from the private sector, and demand from both local and foreign markets, that the aquaculture industry will be able to further contribute to the improvement of the economy of the region. Information related to the fishing community of the Southeast Asia region, their employment, financial status, fishing production, and consumption is limited. Of the 11 countries that comprise the Southeast Asia region, data from ten countries is available from the Our World in Data (OWID) data base [11–13]. Moreover, the relative sluggishness of their economic activity is projected from year 2030 to year 2050 as a result of many major constraints, including fish farming in areas of nearly maximum capacity, limited access to quality freshwater, quality of fish seeds and feeds, competitiveness in land use, volatility in fish prices, challenges to regional governance and regulation, and degradation of the environment [14]. In conjunction with global total fishery production, a better understanding of different patterns in meat, fish, and seafood consumption habits is crucial in investigating the sustainability of food security, as well as the impact of food supply chains at local, national, and global levels [15].

As depicted in Fig. 1, the carbon footprint in Southeast Asia could be quite different from that of other global regions. Part of this variation is due to income differences, the presence of vegetarians or the religious prohibition on eating beef. Fish and shrimp comprise a significant portion of the Southeast Asian population's diet (40% to 50%). As a result, it contributes to more seafood waste production on an annual basis. With reference to data from the Food and Agriculture Organization (FAO), Malaysia is one of the top fish-consuming countries in Asia, with annual consumption exceeding 40 kg/capita since 1980, nearly double the amount of Thailand, Vietnam, Indonesia or Laos, as depicted in Fig. 2.

Meat, fish and seafood consumption per capita between 1970 and 2017

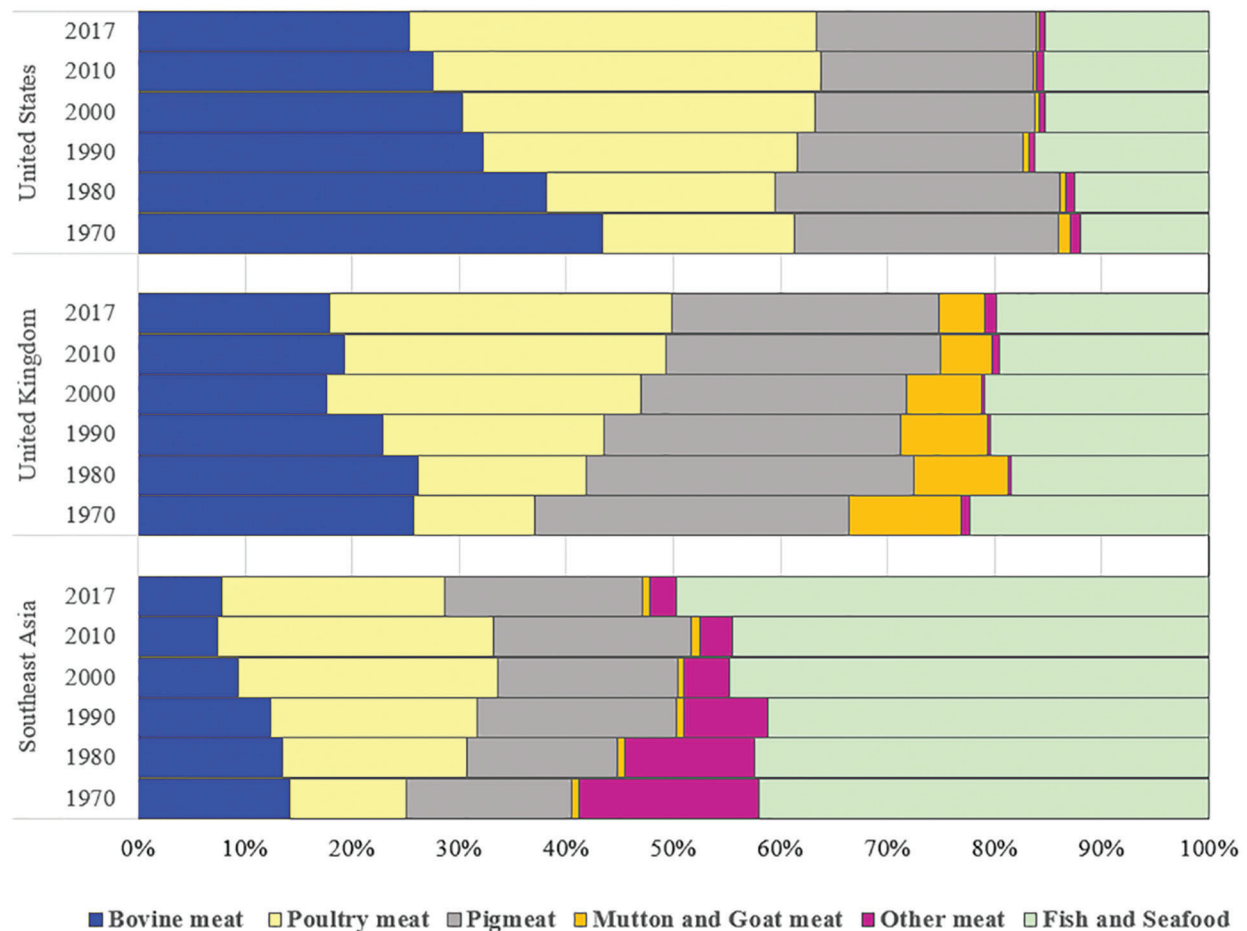


Figure 1: The pattern and amount of meat, fish, and seafood consumption per capita between 1970 and 2017. Fish and seafood are major sources of protein for the population in Southeast Asia [11,12]

The majority of overall fish demand is assigned to domestic consumption, supplemented by exports and non-food use. With increased fish consumption, the overall production of fish and seafood from Southeast Asia has quadrupled in the last 60 years. The inland and marine capture in Southeast Asia during the period 1960 to 2018 has been provided by the FAO, as shown in Fig. 3. Thailand has shown a reduction in fish capture since 2000, whereas the Philippines has shown a tendency to reduce since the 2010s. In Malaysia and Myanmar, fish production has increased steadily since the 1960s, while overfishing in Indonesia could be a concern in Southeast Asia. Increased fishing activity is indicative of humanity's growing footprint in the world's oceans. Over fishing can have a detrimental effect on the fish population.

The exponential growth in seafood production results in approximately 35% to 45% waste, including shell waste, at a volume of about 1 million tons [16]. A large part of this waste is being discharged into the coastal zone, which causes serious environmental pollution. Nevertheless, if properly treated, at least some of the waste can also be a source of interesting raw materials that can have diverse applications [17,18], including composting and recycling into animal feed and organic compost [10,19]. It was also reported that around 35% of food is lost worldwide as a result of spillage and deterioration during storage, processing, distribution, and waste at the consumption point. Food supply chains in developing

nations must be reinforced by among other things, enabling small farmers to organize, diversify, and scale up their production and sales [20]. Bioconversion alternatives, in conjunction with a bio-refinery strategy, have the potential to provide environmentally friendly and cost-effective processes that can support sustainable seafood production [21].

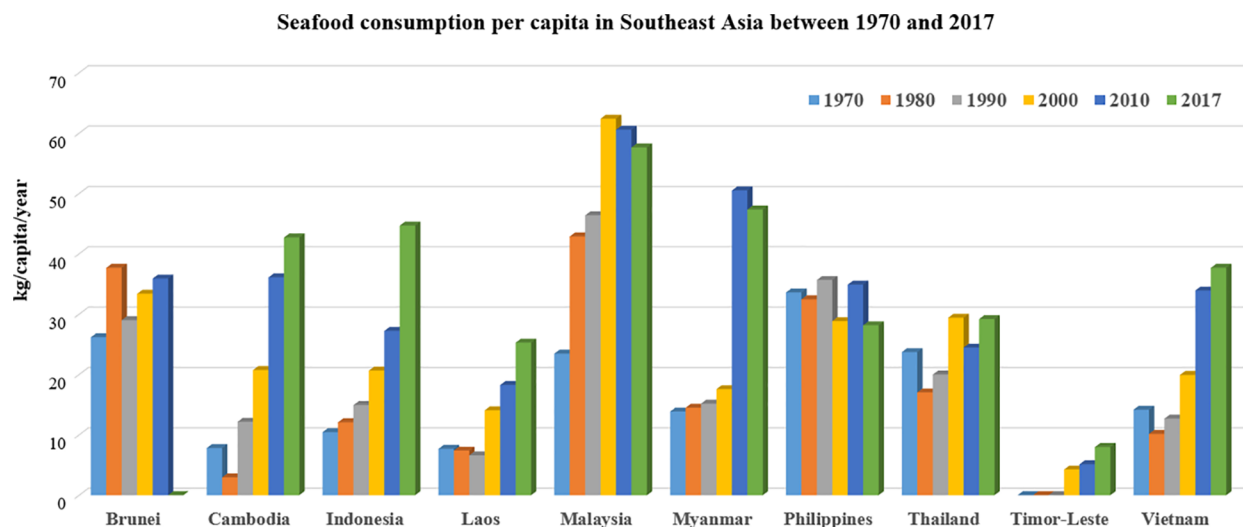


Figure 2: The amount of seafood consumed per capita between 1970 and 2017 in Southeast Asia. Seafood consumption in Malaysia is the highest in Southeast Asia. In general, seafood consumption has increased from the year 1970 to 2017, except in Brunei and the Philippines [11]

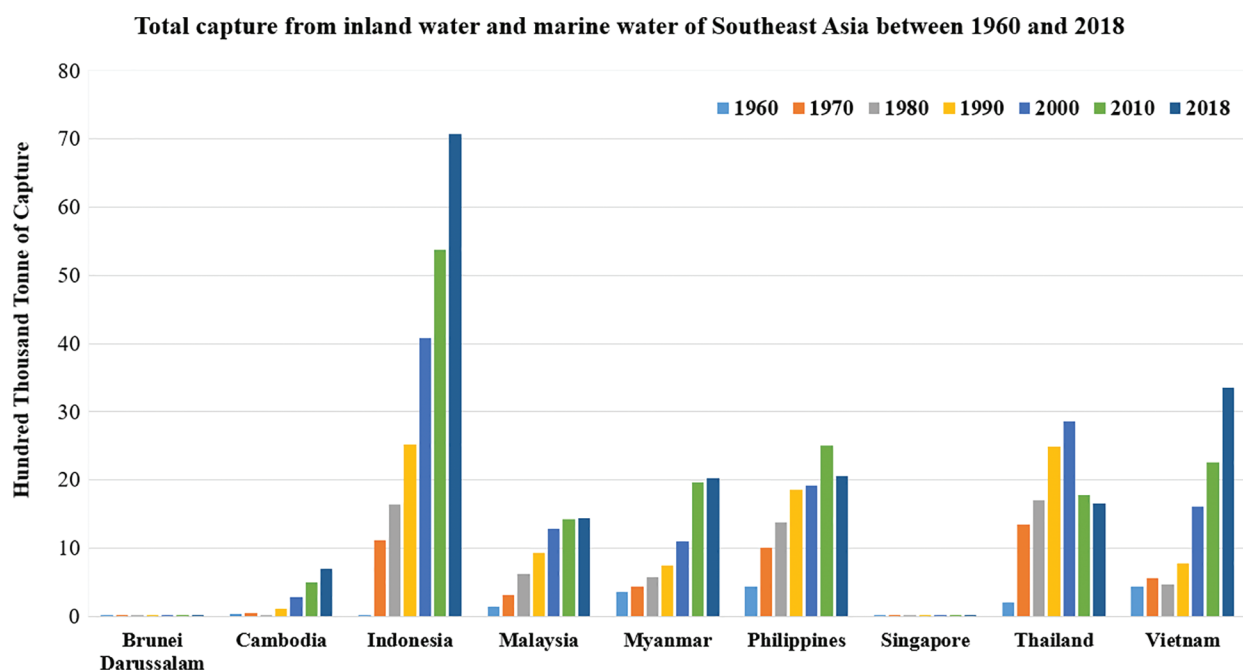


Figure 3: Total capture from inland water and marine water in Southeast Asia between 1960 and 2018. The overall production of crustaceans, diadromous fishes, freshwater fishes and mollusks from capture has been increased since the 1960s in Southeast Asian countries [11,12]

In Southeast Asia, total seafood capture and aquaculture production are 183 and 128 hundred thousand tons respectively. Meanwhile, in 2017, fish and seafood consumption was 320 kg/capita/year [3]. In this context, tons of chitosan are potentially produced annually, especially from shrimp, fish scales, and crab shell residues. Indeed, increased fishing, aquaculture and seafood consumption can provide large quantities of chitin and chitosan as biowaste [22] and can be extracted commercially as competent polymers. However, current waste management in Southeast Asia is insufficient, and has negative environmental and human health consequences. The contamination of landfills in Southeast Asia is getting serious. The rate of waste generation is associated with the different levels of income and the degree of urbanization as well. Many landfill sites are unsanitary as a result of land acquisition issues, poor collection and disposal fees, and an insufficient number of landfill sites. While there are a range of laws relating to waste management, however people do not honor the waste management laws in most countries. In some underdeveloped countries, the laws on waste management are especially lacking. Most ASEAN countries are uncertain about the institutional structure for waste management. Moreover, they lack of resources, especially funding, technology, waste management, capabilities, and skills [8,23,24]. Therefore, it is important to properly monitor the production, storage, collection, transfer, transportation, transformation, and disposal of shellfish waste. Strict rules on the handling of waste shells should be applied.

3 Crustacean Waste

Crustacean shells are typically made up of 20% to 50% calcium and magnesium carbonate, 20% to 40% protein, and 15% to 40% chitin [25]. Chitin and its main derivative, chitosan, are promising biomaterials, that are natural aminopolysaccharide polymers. These materials are crucial in strengthening and protecting marine crustaceans, mollusks, insects, shrimp, and fishes [26]. Chitin is closely related to ingredients like protein, calcium carbonate, and lipids. Protein can be composted and used as animal feed, whereas calcium carbonate is used as a nutritional supplement, an antacid, and in industries such as plastics and construction [27].

It is estimated that Southeast Asia will generate one quarter of the world's seafood by 2030, and that aquaculture, fisheries and shrimp waste will produce biological waste, and cause a range of environmental problems. Many shrimp species in Southeast Asia are prone to diseases. Currently, only a small number of the Penaeidae family are farmed. Among shrimp species cultured in Southeast Asia, the genera *Metapenaeus* (14 species) and *Parapenaeopsis* (9 species) are the common predominant species [28,29]. The members of the Penaeidae are all marine, although they are mostly found in coastal and swampy areas where the water is typically of a high salinity brackish type. However, for the genus *Metapenaeus*, they breed fairly far off at sea. Generally, the larval stages shift towards the coastal areas in order to grow up and return to the offshore grounds at maturity to breed [30]. On the other hand, the members of the Caridea are primarily freshwater, in particular the members of the *Macrobrachium* genus. There are some exceptions, *Macrobrachium equidens*, *Macrobrachium rosenbergii*, *Caridina propinqua*, *Palaemon styliferus*, and *Palaemon tenuipes* are also found in brackish water [31,32].

The output of shrimp waste from the processing industries has greatly increased in recent years as shrimp farms and catches have grown. Indeed, 40% of the shrimp are edible, and the remainder is discarded, accounting for 45% to 50% of the wet weight [33,34]. Shrimp shells are the most common commercial source of chitin, which is used to make chitosan, chito-oligosaccharide, and glucosamine [35,36]. Generally, chitin encrusted with calcium carbonate, collagen, astaxanthin, and lipid residues make up about 20% to 40% of shrimp biowaste [37,38]. Astaxanthin is in high demand for food, feed, nutraceutical, and medicinal applications. This has prompted major efforts to increase the supply of astaxanthin from biological products or biowaste rather than synthetic sources. Therefore, recently, the production of astaxanthin from natural sources has emerged as one of the most exciting biotechnological practices [39].

Decaying crab waste can release ammonia gas and nitrates, and can potentially contaminate ground water and freshwater wells that provide drinking water. Waste recycling could reduce the risk of crab shell deterioration in the environment [40,41].

Advances in commercialization of the source of chitin could reduce the overall cost of bioremediation technology. It is able to adsorb metal ions and also hydrophobic organic molecules, making it valuable in wastewater treatment and other industrial applications [42]. Volatile fatty acids, alcohols, and ammonia have been shown to be involved in chitin fermentation [43]. Chitin from crab waste contains nitrogen and is commonly used in pharmaceutical, carbon dioxide or textile industries, as opposed to many other biomass forms [8], and is used for emulsifying food additives. It is also marketed as an anti-inflammatory supplement to reduce cholesterol, encourage weight loss, and regulate blood pressure. Chitosan can be used to manufacture biodegradable plastic [44].

Indeed, better collaboration and investment by entrepreneurs and environmental officials are helpful to mitigate waste disposal problems [45]. Chitin and its deacetylated form chitosan are biomaterials of great importance with various roles in different fields. A description of the amount of chitin from different waste resources from Southeast Asia is summarized in Table 1. The percentage of chitin from shrimp shell waste is between 13.5% to 43.8% [46,47], 34% to 37% chitin available from squid shell waste [48]. Recoverable amounts of chitin from crabs and lobsters are between 16% to 20% [49–51]. A summary of the application of organic compounds from shrimp shells and crab shells is tabulated in Table 2.

Table 1: Proximate elemental composition of common crustaceans available in Southeast Asia

Sources	Protein (%)	Chitin (%)	Ash (%)	Lipid (%)	Species	References
Shrimp shells	44.0	18.0	22.8	7.3	<i>Metapenaeus affinis</i>	[46]
	49.6	13.5	21.9	6.3	<i>Penaeus monodon</i>	[46]
	39.4	19.9	32.0	3.8	<i>Xiphopenaeus kroyeri</i>	[46]
	25.9	43.8	27.9	2.4	<i>Litopenaeus vannamei</i> (abdominal)	[47]
Crabs	40.7	23.2	22.2	13.9	<i>Litopenaeus vannamei</i> (head)	[47]
	25.98	18.83	36.34	4.54	<i>Portunus pelagicus</i>	[49]
	n.a.	16.8	37.7	n.a.	<i>Scylla serrata</i>	[50]
Lobsters	26.0	20.4	38.2	3.0	<i>Thenus orientalis</i>	[49]
	14.0	16.0	54.0	1.1	<i>Panulirus argus</i>	[51]
Squids	57.2	34–37	1.37	1.9	<i>Loligo chensis</i>	[48]

Note: n.a. not available.

Table 2: Potential application of organic components from shrimp shells and crab shells

Sources	Chemical compounds	Potential applications	References
Crab	Calcium carbonate (40%–70%)	Conversion to calcium hydroxyapatite, inhibits demineralization of the teeth. Pharmaceutical, agricultural, construction and paper making industries. Coagulant for tofu making. Adsorbent for phosphate removal. Defluoridation of drinking water.	[8,52–55]

(Continued)

Table 2 (continued)

Sources	Chemical compounds	Potential applications	References
	Chitin (15%–40%)	Food supplements, food and feed additives, pharmaceutical products, metal ligands, groundwater bio-denitrification, fertilizer, textile paper, film and sponge making industry, enzyme immobilization, biomedical and wound dressing, blood anticoagulant, antithrombogenic and hemostatic materials, dentifrice, dental materials, tissue engineering, cosmetics and cosmeceuticals.	[8,56–62]
	Protein (13%–40% of dry weight)	Fertilizer, animal feed and pet feed.	[63–65]
	Carotenoid (34.4 g/100 g marine crab; 35.7 g/100 g freshwater crab)	Aquaculture feed.	[66–68]
	Astaxanthin (23.6 g/100 g marine crab; 7.2 g/100 g freshwater crab)		
	Lipid (0.37%–0.65% of total lipids)	Nutritional aquaculture feed. Potential application of phospholipids, anticancer and anti-inflammation.	[53,69–71]
Shrimp	Calcium carbonate (40%)	Plant fertilizers, calcium oxide nano-plates, animal feed, pharmaceutical products, defluoridation of drinking water, and tooth remineralization materials.	[55,72–75]
	Chitin (14%–30%)	Antibacterial finishing of bamboo rayon, water treatment through adsorption of heavy metals, aquaculture feed with growth promoter and immuno-stimulant, dental resin sealant, dentifrices, wound healing, enzyme immobilization, textile industry.	[76–83]
	Protein (33%–40%)	Animal feed and aquaculture feed.	[65,84]
	Astaxanthin (14–39 mg/kg)	Dye, drug or food supplement.	[33,65,85]
	Lipid (0.3%–10%)	Aquaculture diets.	[65,67,86]

4 Characteristic of Chitin and Chitosan

Chitin is a highly insoluble linear polysaccharide $(C_8H_{13}O_5N)_n$, with repetition units of N-acetylglucosamine (GlcNAc), which is a monosaccharide derivative of glucose, as depicted in Fig. 4c. These

units form covalent β -1,4 linkages. This natural polysaccharide was first discovered in 1884 [87], the second most abundant natural polymer, only second to cellulose [35]. These polymer molecules are versatile and can form solid structures on their own or combine with other components such as calcium carbonate and metal silicates to make even stronger substances [88]. Chitin has its own unique assembling manner naturally. It can be assembled in at least three different manners, and these are allomorphs named as α -, β -, or γ -chitin. α -Chitin is composed of anti-parallel chains of GlcNAc. β -Chitin, is a less common allomorph that contains chains aligned in a parallel manner, which results in weaker intermolecular interactions as compared to the α -Chitin. Furthermore, γ - has two chains that run parallel to each other and a third chain that runs antiparallel to the two [35,89,90]. Other than the chain assemble manner, the physicochemical properties such as X-ray diffraction and thermal properties of the α -, β -, and γ - chitin are very different. Jang et al. [91] reported that α -chitin has 4 crystalline reflections, β -chitin has 2 crystalline reflections, and γ -chitin shows a combination of the characteristics of α - and β -chitin.

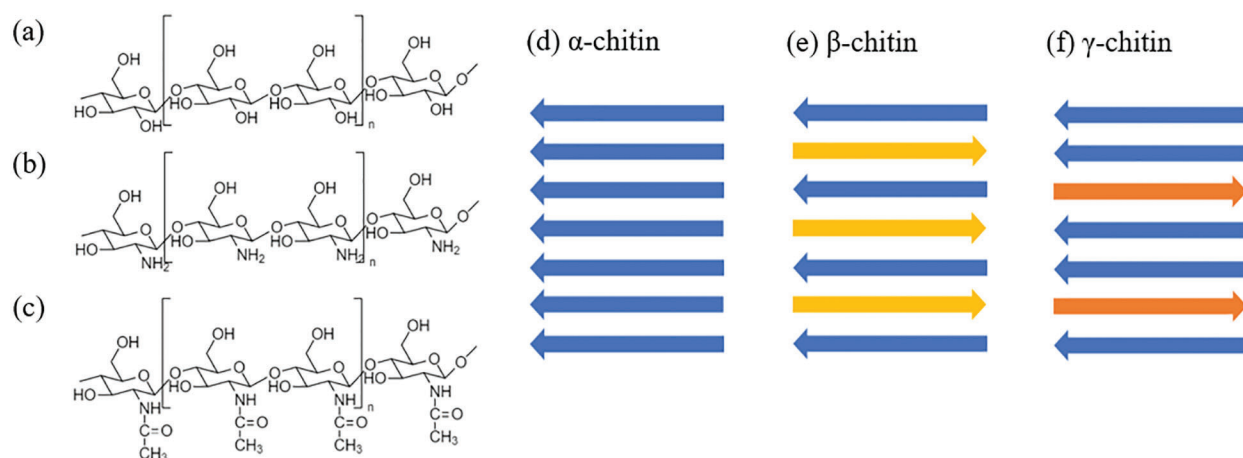


Figure 4: The molecular structure of (a) cellulose, (b) chitosan and (c) chitin (n = repeating unit), and different chitin conformations (d) α -chitin, (e) β -chitin and (f) γ -chitin

Chitosan is a type of N-deacetylated chitin that is made up of a random distribution of β -(1–4)-linked D-glucosamine and N-acetyl-D-glucosamine. Chitin deacetylation can be accomplished in alkaline circumstances by utilizing a higher percentage of alkaline solution or enzymatic treatment with chitin deacetylase [90]. Chitosan deacetylation typically ranges from 50% to 98%, with commercially available chitosan deacetylated at an average of 80% [90,92,93]. Chitosan has certain limitations in its pure form, some chemical modifications can improve its chemical properties. On the other hand, chitosan can be soluble in acidic aqueous solutions depending on the degree of deacetylation [35,87]. Water solubility increases with the increasing degree of deacetylation [90].

Chitin and chitosan have various applications and advantages, as depicted in Fig. 5. Biodegradability, nontoxicity, biocompatibility [94,95], and mucoadhesive properties [96] are the most desirable properties of chitosan. Chitosan in various forms can be used, including powder [97], flakes [98], fibers [99], composites [100], films or membranes [101]. Chitosan can be converted into nanoparticles or nanofibers [102,103]. Chitin is commercially important in comparison with synthetically replaced cellulose (1.25%) due to its high concentration of nitrogen, which is 6.89% [104]. Chitin is a valuable agent for chelating. Chitosan is nonetheless more suited for several fields, including farming, waste water therapy, dental science, cosmetics, pharmaceuticals, and the food industry, compared with the chitin sector [88,105,106]. Chitin derivatives are interesting biopolymers that can increase the viscosity of vegetable oil for

bio-lubricant production [107]. The physical-chemical features of shrimp shell chitin and its function in the development of arthropod exoskeletons are being investigated to explore chitin asthma and allergy [108]. Chitin is used as a substrate for chitin deacetylase [109].

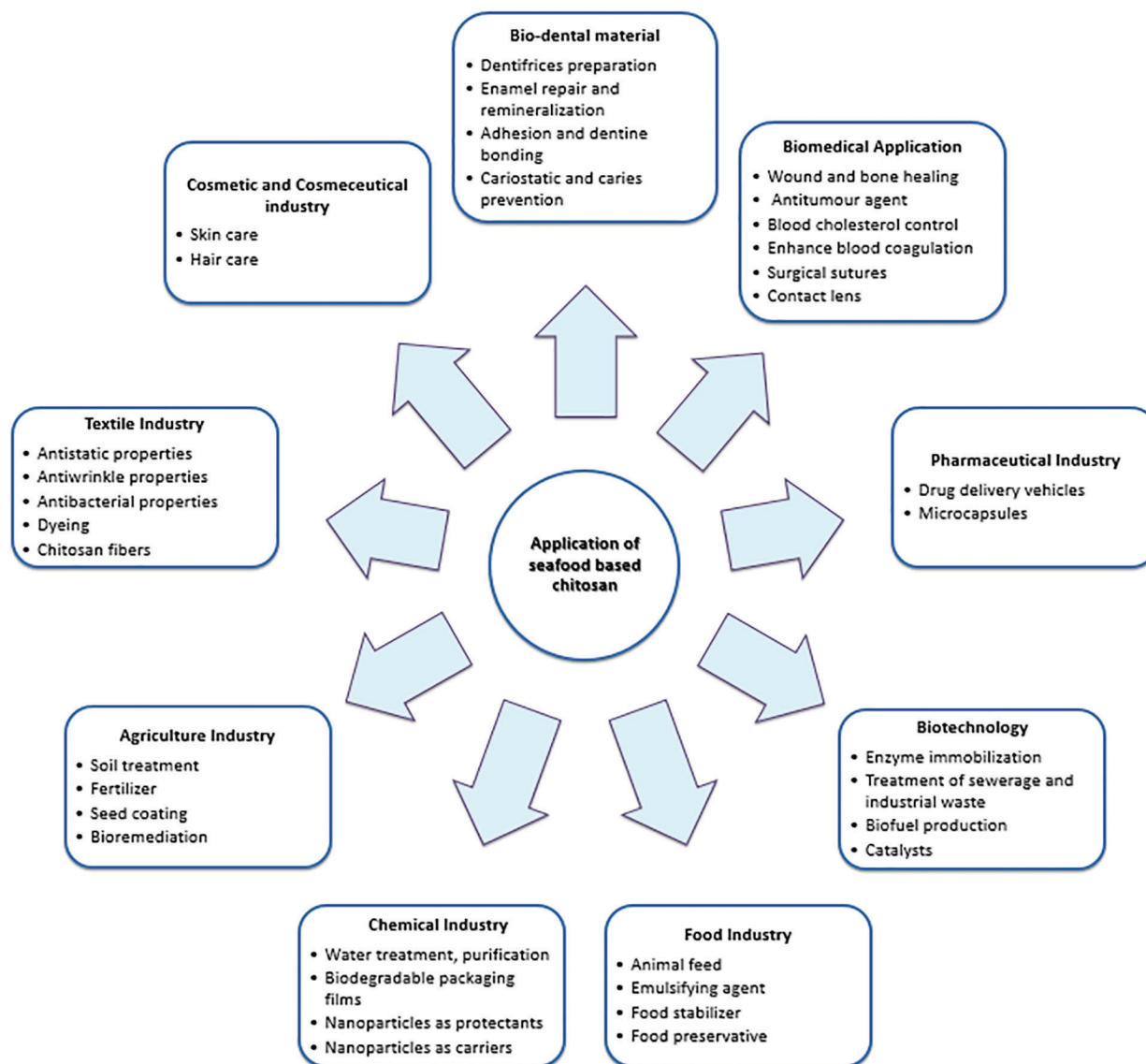


Figure 5: Applications of seafood based chitosan in various industries [61,110–112]

5 Techno-Economic of Chitosan Production

The processing and generation of crustacean waste in Southeast Asia are rising at a fast rate. Chitosan from both crustacean and fish wastes are obtainable by different techniques, and is based on the composition of the source or organisms. These nutraceutical compounds are: chitin, chitosan, glucosamine, astaxanthin, and shrimp carotenoprotein. For the time being, the majority of these approaches rely on chemical and physical agents for mineral, protein, and pigment removal, as well as deacetylation [113,114]. The first step in conventional separation of chitin from waste is mineral removal from the fundamental structure of the exoskeleton. The following step is the removal of proteins associated with all components. The order

of deproteinization and demineralization by mineral acid is known to be interchangeable. The third step is the removal of pigments such as carotenoid [115].

The objective of demineralization is to remove calcium carbonate, calcium phosphate, and other minerals by using dilute acids at room temperature, with reaction times varying from 1 h to 24 h [116,117]. Higher acid and basic concentrations are the first way of obtaining chitosan. The demineralization could take place under 10% hydrochloric acid (HCl), room temperature, a solid/solvent ratio of 1:10 (w/v), and within 22 h of stirring conditions. At 70°C, at a solid/solvent ratio of 1:15 (w/v), the desalted shell would be decomposed into 10% sodium hydroxide (NaOH) over 24 h [116,117]. In general, chemical extraction has an inevitably adverse effect on the molecular weight and degree of acetylation. This is most likely due to the rupture of the chitosan polymeric chain in the presence of a concentrated hot NaOH reaction over a long incubation time. The length of the chitosan molecule determines its viscosity. Therefore, in order to yield very viscous chitosan in a short incubation period, increase the reaction temperature by 10°C or increase the incubation time by one hour [118].

Lertsutthiwong et al. [119] extracted chitin from shrimp through chemical treatment. In the first attempt, deproteinization treatment with NaOH (1%, 2%, and 4%) was performed and followed by demineralization using 4% HCl, whereas in the second attempt, demineralization was performed and followed by deproteinization. The result revealed that, if deproteinization precedes demineralization, chitin becomes whiter with an increased concentration of NaOH. With the usage of 4% HCl, the process of decalcification would be adequate to extract minerals within 2 h. Indeed, prolonged decalcification time would not cause much difference in the content of ash, but it might contribute to polymer degradation. Once chemical treatment begins with 4% HCl demineralization for either 2 h or 12 h, colored matter and protein will remain bound to the solid matrix. These products should be separated during subsequent deproteinization. An increase in the proportion of NaOH would enhance the removal of protein and colored materials. The yield of chitin for most conditions was 20% to 27% [119].

When deproteinization is first performed, the defensive protein coating is removed and the unprotected chitin is exposed to HCl. This causes effective demineralization as well as increased hydrolysis and probable material loss in the solid chitin fraction. On the other hand, if demineralization is achieved first, chitin is covered by adhering collagen, resulting in less backbone chain hydrolysis and a higher yield of chitin [119]. Therefore, demineralization should also be followed by deproteinization for the development of high viscosity chitosan. Over the years, many methods for producing chitosan from various seafood wastes have been produced and proposed by several researchers. All of these formed the basis for the development of the chitosan industry by chemical process. However, the majority of the treatments indicated have been carried out at high pressure temperatures above 100°C.

Generally, proteins are removed with mild alkaline solutions (1% to 10%) ranging from room temperature to boiling temperature, reaction time varies from 0.5 h to 24 h. An increase in the sample-to-solution ratio above 1:4 (w/v) only has a small impact on chitosan quality [120]. In the year 2005, Naznin [121] performed both demineralization and deproteinization with different concentrations of NaCl (10% to 50%) and NaOH (0.5, 1.0, 1.5, and 2.0 N) at room temperature. The best combination was overnight demineralization with 30% HCl followed by deproteinization with 1.5 N NaOH. Puvvada et al. [122] deproteinized samples by boiling them in 2% and 4% NaOH for 24 h, followed by demineralization in 1% NaCl. The chitosan yield was found to be 35.49%. Hossain et al. [123] demonstrated that 3% HCl and 4% NaOH at ambient temperatures ($28 \pm 2^\circ\text{C}$) were appropriate for demineralization and deproteinization. In order to avoid undesired modification of native chitins, ethylenediamine-tetra-acetic acid (EDTA) can be added for demineralization [124,125]. In general, the yield of chitosan is promising with chemical approaches, but a major drawback is that a great deal of water is used to extract the target component and to neutralize acid.

Recently, Alshehri et al. [126] applied some other steps in the production of chitosan. Along with the protein removal procedure, the researchers utilized oven-dried shrimp shells and boiled them for 1 h in a solution containing 2% and 4% w/v NaOH. The treated shrimp shells were pulverized to produce sizes ranging from 0.5 mm to 5.0 mm. The mineral removal protocol was accompanied by 1% HCl and 4 parts of samples for 24 h. The dried shrimp shell powder was treated with 50 ml of 2% NaOH solution for about 1 h to decompose albumin into water-soluble fractions. During the deacetylation process, the obtained chitin was boiled in 50% NaOH for 2 h at 100°C. The sample was repeatedly washed with 50% NaOH. Eventually, the sample was dried in a hot air oven at 110°C for 6 h to obtain a milky white powder.

For the bleaching procedure, the collected precipitate will be treated for 1 h with 1% potassium permanganate solution and then with 1% oxalic acid solution for another 1 h. Finally, at the boiling temperature, the sample is treated with an alkaline solution. The deacetylation (DDA) and molecular weight of chitosan are important for assessing the nature of the physical and biological properties of chitosan. Chitin deacetylation allows 40% to 45% NaOH at a temperature of 60°C to 130°C to remove acetyl from chitin for 12 h to manufacture chitosan [35,127]. From the literature, two main methods have been proposed, including heterogeneous and homogeneous settings. Chitosan formed by N-deacetylation of chitin under uneven conditions has a normal distribution of GlcNAc blocks, whereas chitosan prepared under homogeneous conditions has coalescence of both GlcNAc and GlcN random (Berneuil) [128]. An important property used to characterize chitosan is its solubility. According to this parameter, chitosan is sufficiently deacetylated with chitin to form a dilute hydrochloric acid soluble salt (DDA 80% to 85% or more), with an abundance of amino groups in the molecule, and can be considered as cationic polyelectrolytes [129].

Another green chemical method that has recently emerged is the selective extraction of chitosan via deep eutectic solvent (DES). DES is the greener alternative to conventional ionic liquids, fluids that can be self-associated with hydrogen bonding interactions in order to form eutectics with lower melting points compared with each individual component. DES is the appropriate mixture of hydrogen bond acceptor and donor. Ionic liquids are a singular form of liquid consisting of a very large organic cation and inorganic or an organic anion that melt below 100°C. The advantages of ionic liquids, as indicated by DES, are generally safe for the environment, including low toxicity, low vapor pressure, chemical and thermal resilience, non-flammable, potential high solubility, fusion, and low adaptability [130,131]. In addition, treatments are less expensive and use less water. This approach also shows great potential to overcome the difficulties and processing capabilities of difficult-to-dissolve bio-fertilizers such as cellulose, lignin, and starch [132].

This approach for chitosan production, including the application of ionic liquids, has been noticed as early as 2001. The design of DES was first presented by Abbott et al. [133], and DES was classified into four groups, type I being a combination of organic salts and metal salts, type II being a combination of organic salts and metal hydrate, type III is a mixture of organic salts and a compound that is a hydrogen bond donor, and type IV is a combination of a metal chloride and a compound that is a hydrogen bond donor [134]. As per depicted in Table 3, potential chemical combinations in the proper ratio enable chitosan extraction from seafood waste. When blended in the correct ratio, many primary metabolites, such as sucrose, glucose, mannose, fructose, citric acid, malonic acid, maleic acid, and natural organic acids from plants, will change their state from solid to liquid. This discovery resulted in a new class of DES called natural deep eutectic solvents (NADES). Nevertheless, modern ionic liquids and deep eutectic solvents do have drawbacks to be introduced to the current chemical industry. Operating temperatures above 90°C increase maintenance costs, and there is a risk of human and environmental pollution due to the lack of solvent recycling [135,136].

The green chemistry process was further improved by Feng et al. [137], they transformed shrimp shells into o-attached chitin directly using NADES, however the working temperature was above 90°C. Meanwhile, the one pot process of decalcifying, deproteinization, and acylation is able to remove the minerals and proteins simultaneously. In 2020, Bradić et al. [136] demonstrated chitin extraction by using NADES at a lower temperature of 60°C to 90°C. In addition, these same NADES can be recycled without loss in the capacity to break the shrimp body. The transformation from traditional chemical extraction to a green chemical approach provides a simple, efficient, and environmentally friendly approach to producing value-added chitin derivatives from seafood waste.

Table 3: Chitin dissolution by using choline chloride as the hydrogen bond acceptor (HBA)

Waste	Hydrogen bond donor (HBD)	Ratio (HBA: HBD)	Reaction temperature (°C)	Reaction hours	References
Lobster cells	Thiourea	1:1	90	2	[138]
	Urea	1:2	90	6	[138]
	Glycerol	1:2	100	4	[138]
	Malonic acid	1:2	100	2	[34]
Crab shells	Thiourea	1:2	100	6	[34]
	Urea	1:2	100	10	[34]
Shrimp shells	Lactic acid (85%)	1:1	60	6	[136]
	Malonic acid	1:1	60	6	[136]
	Urea	1:2	80	6	[136]
	Citric acid	1:1	80	6	[136]

5.1 Importance of Biological Extraction

Common chemical extraction methods for chitin often require varying combinations of demineralization, deproteinization, decoloration, and deacetylation procedures. The use of strong acid and alkali solutions at high temperatures or pressures with lengthy incubation is the major drawback of these chemical processes. The entire process consumes a lot of energy as well as a lot of acid or base. Furthermore, chemical processing can harm the physicochemical characteristics of chitin and result in chemical-containing effluent discharge. Despite the use of deep eutectic solvents, which may be an alternative green chemistry application, the use of urea, thiourea, choline chloride, and other chemical compounds may have a negative impact on the environment, low toxicity, and limited biodegradability. Therefore, the expenses, the use of electricity, and higher risk in the production and purification process could restrict industrial chitosan extraction [139].

Despite the fact that the biological extraction process is now limited to laboratory experiments, it has the potential to replace chemical techniques that have a number of commercial limitations. Further advancements in minimizing chitin degradation and lowering contaminants to acceptable levels may boost the biological extraction advantage of chitin. In order to overcome these concerns, a biotechnology approach has been proposed, the use of proteolytic enzymes in digesting proteins or the use of microorganism-mediated fermentation [140,141]. The focus of chitosan manufacture by biological extraction is on the discovery of novel and improved production processes. Enzymatic conversion, chitosan biological activity networks, and chitosan physicochemical properties after microorganism-mediated fermentation [142].

5.1.1 Chitin Enzymatic Deproteinization

Many literature recommends the replacement of chemical deproteinization with enzymatic methods or microbial proteases. Various proteolytic enzymes, including alkalase, pancreatin, papain, pepsin, and trypsin, were utilized to extract protein from chitin. Enzyme detection and operational state optimization are vital, and 5% to 10% of the residual protein remaining coupled to purified chitin may be a less efficient enzyme process. The final, isolated chitin is next treated with moderate alkaline treatment for a brief duration, to increase its purity and retain the structure of chitin [87]. The order of demineralization and deproteinization in biological extraction is mostly irrelevant to the yield and uniformity of chitin purification. Minerals can obstruct proteolytic enzymes from accessing the substrate. As a result, demineralization will be conducted first to increase tissue permeability, limit the interaction of potential enzyme inhibitors, and promote proteolytic enzyme activity.

Younes et al. [87] improved enzyme deproteinization before demineralization. Several microbial proteases were compared in this study based on their efficacy in shrimp shell deproteinization. *Bacillus mojavensis* A21, *Bacillus subtilis* A26, *Aspergillus clavatus* ES1, *Bacillus licheniformis* NH1, *Bacillus licheniformis* MP1, and *Vibrio metschnikovii* J1 were applied for enzymatic deproteinization. Using an enzyme to substrate (E:S) ratio of 7.75 U/mg, 60°C, and 6 h of incubation, *Bacillus mojavensis* A21 protease can remove up to 77.3% of shell proteins. Chitin samples were then characterized and compared to the samples prepared by chemical deproteinization [143]. Hamdi et al. [144] isolated chitin from *Portunus segnis* blue crab and *Penaeus kerathurus* shrimp using chemical demineralization and enzymatic deproteinization approaches. For blue crab shells, the percentage of deproteinization was about 85% and for shrimp was about 91%, with an E:S ratio of 5 U/mg after 3 h of incubation at 50°C. The use of enzymatic deproteinization also limits the cost of commercially available enzymes. Due to the current availability of coexisting proteases, crude extracted proteases derived predominantly from bacteria from fish viscera are less expensive and more effective [87]. Recently, da Silva Lucas et al. [145] isolated chitin and chitosan from the exoskeletons of edible insects. Enzymatic hydrolysis began with the addition of the enzyme alcalase in a 2% (w/w; E:S) proportion, followed by 8 h of deacetylation with 40% NaOH at 90°C with continuous mechanical agitation.

5.1.2 Extraction by Fermentation

Another cost-effective method for chitin extraction from seafood waste involves two separate fermentation processes, one with and one without lactic acid bacteria. This method consists of fermenting chitin sources in a controlled fermentation phase with microorganisms. The process can be accomplished by the addition of selected microbe strains, one- and two-stage fermentation, co-fermentation or subsequent fermentation [146]. Lactic acid bacteria fermentation is a revolutionary approach to enzymatic chitin extraction and can be combined with chemical treatments that reduce the amount of acid and alkali required. The bacteria strains ferment waste materials and produce organic acids *in situ*. *Lactobacillus* sp. strains, particularly *Lactobacillus plantarum*, *Lactobacillus paracasei*, and *Lactobacillus helveticus*, are the most commonly utilized bacterial strains for fermentation. *In situ*, the bacteria strains ferment waste materials and create organic acids [147,148].

During the fermentation process, two phases will be formed. Liquid fraction rich in proteins, minerals and pigments, and a solid phase containing rudimentary chitin. The action of lactic acid will facilitate the precipitation of chitin and the formation of calcium lactate after chemical reaction with calcium carbonate. Lactic acid obtained by glucose conversion concurrently helps to lower the pH level and thus activates proteases. This method was also utilized to extract additional components from silage shrimp waste, such as carotenoids [147,148]. The shrimp and crab shells contain higher amount of calcium, in general *Lactobacillus plantarum* is efficiently used to remove calcium after deproteinization [149].

Bacillus subtilis [150], *Pseudomonas aeruginosa* [151], and *Serratia marcescens* [152,153] also been reported as effective types of bacterial strains for protein digestion.

Khanafari et al. [154] reported a comparative analysis of the extraction of chitin and chitosan from shrimp waste by chemical and microbial methods. The biochemical process of extraction involved deproteinization with 2% w/v NaOH solution (30:1 v/w, 90°C, 2 h), separation of alkali-insoluble fraction by centrifugation (4000 rpm, 15 min), extraction of chitosan from the fraction under reflux (10% v/v acetic acid 40:1 v/w, 60°C, 6 h), followed by separation of crude chitin by centrifugation (4000 rpm, 15 min), and precipitation of chitosan from the extract at pH 9, which adjusted with a 4 M NaOH solution. The yield of chitin and chitosan by alkali-acid treatment were 510 mg/g and 410 mg/g, respectively. On the other hand, biological extraction of chitin by using *Lactobacillus plantarum* (PTTC 1058), *Lactobacillus acidophilus* (PTTC 1643), and *Lactobacillus rhamnosus* (PTCC 1637) provides high reproducibility in lesser time, decreased solvent consumption, lower energy intake, and is environmentally friendly since no waste is produced. The *Lactobacillus plantarum* (PTTC 1058) was reported as an attractive source of recovery for chitin and chitosan, chitin and chitosan yields were 700 mg/g and 420 mg/g, respectively [154]. The findings showed that the microbial approach was superior to the chemical method.

Duan et al. [155] investigated the fermentation of shrimp waste on a laboratory and pilot scale with *Lactobacillus acidophilus* for 48 h of fermentation. In pilot fermentation, the pH was 3.99 and 3.86, respectively, after 12 h and 24 h of fermentation. The mineral and protein content were 0.98% and 8.44% respectively after 48 h of fermentation. The remainder of fermented shrimp waste consists of less than 1% minerals and is rapidly turned into chitin with a simple bleaching process. Recent advancements in fermentation developments have led to the practical deployment of diverse manufacturing procedures to replace old chemical synthesis methods. In industrial applications, these natural processes may be exploited to generate the final product in a regulated manner. For the processing of bulk chemicals and enzymes, solid state fermentation offers several advantages. *Lactobacillus* sp. and *Clostridium* sp. are two popular bacterial strains utilized for ensilation [148]. The ensilation can be used in addition to organic acids and *Lactobacillus* sp. treatment. Shrimp waste ensilation can be accomplished biologically or chemically. The ensilation of crustacean shells can be accomplished through *in situ* processing of low-cost organic acids, the production of lactic acid by bacteria, which results in the liquefaction of semi-solid waste, a low pH, and the activation of proteases. The process provides numerous advantages as a method of preservation and allows for the recovery of value-added by-products such as chitin, proteins, and pigments to a broader market [84,156,157].

Pseudomonas sp., *Bacillus* sp., and *Aspergillus* sp. have been considered for chitin recovery. Sedaghat et al. [158] reported the recovery of chitin from shrimp shell waste by utilizing protease secreted by *Pseudomonas aeruginosa*. The influence of three parameters on the demineralization and deproteinization of shrimp shell waste was investigated, including glucose concentrations, inoculation levels, and fermentation period. According to the findings of this study, the ideal conditions for chitin extraction are 20% glucose, 20% inoculum, and 6 days of fermentation, which resulted in 82% and 94% demineralization and deprotenization, respectively. *Bacillus pumilus* A1, *Bacillus mojavensis* A21, *Bacillus licheniformis* RP1, *Bacillus cereus* SV1, *Bacillus amyloliquefaciens* An6, and *Bacillus subtilis* A26 were examined by Ghorbel-Bellaaj et al. [159] for fermentation of shrimp waste. The data demonstrated that all *Bacillus* strains could deproteinate shrimp waste. The highest degree of deproteinization was attained by *Bacillus cereus* SV1. Teng et al. [160] conducted an interesting study on the continuous generation of chitin from shrimp shells, engaging *Aspergillus niger* for fermentation. The fungal proteolytic enzymes deproteinate and demineralize shrimp shells, releasing amino acids that provide nitrogen for fungal growth.

During bacteria fermentation, both demineralization and deproteinization occur mostly at the same time, lactic acid bacteria produce lactic acid. Microorganisms mediated by deproteinization and demineralization can produce a fraction of liquor rich in proteins, minerals, and carotenoids, particularly astaxanthin and a chitin-rich solid fraction. This fraction may be used as feed for animals [84]. The biologically isolated chitin from shrimp head debris was co-fermented with the protease-producing bacteria *Bacillus licheniformis* 21,886 and the acid-producing bacteria *Gluconobacter oxydans* in a series of steps. The deproteinization and demineralization efficiencies were 87% and 93.5%, respectively, when waste was initially co-fermented, at a chitin value of 90.8% [161].

Zhang et al. [162] conducted a large-scale, three-step fermentation of *Serratia marcescens*, *Lactobacillus plantarum*, and *Rhizopus japonicus* M193 for chitin extract. For the processing of chitosan in normal consistency, 1 M NaOH and 1 M HCl were used to extract fermented residues at room temperature for 24 h and the resulting chitosan solids were dissolved in 2% acetic acid at 30°C for 16 h and centrifuged at 12,000 rpm for 20 min, followed by precipitation by 40% NaOH up to pH 9.0 and then separated by centrifugation at 12,000 rpm for 20 min. The efficiency of both deproteinization and demineralization utilizing *Lactobacillus plantarum* could be achieved during the first two days. On the second day, the pH of the slurry would drop even further. The formation of lactic acid is inextricably connected to continuous demineralization. Simultaneous enzymatic hydrolysis and fermentation was also reported by Dun et al. [163], chitin extraction from crayfish shell waste powder was investigated by using *Bacillus coagulans* LA204, and proteinase K. This one pot reaction was performed at 50°C with 5% (w/v) shell waste, 5% (w/v) glucose, 1000 U proteinase kg⁻¹ shell waste, and 10% inoculation of *Bacillus coagulans* LA204.

In particular, a single-step extraction of chitin may be used by co-culture of lactic acid-producing bacteria and proteases. The solubilized proteins and minerals derived from the fermentation procedure can be processed into animal feed. The optimized green techniques have many beneficial effects and efficiency on recycling seafood waste, water usage could be further reduced to 50% in comparison to conventional methods. Another advantage of applying microorganism in chitosan extraction from seafood waste is the reduction of acidic and alkaline solutions usage. The production costs would be further reduced. The seafood waste as raw materials, implying a new component in the crustacean production chain that includes the fishing, aquaculture, and processing industries. Biologic fermentation technologies have been envisioned as one of the most ecologically sustainable, safe, technologically scalable, and commercially feasible alternative techniques [154].

5.1.3 Biological Deacetylation

The deacetylation procedure is a hydrolysis of acetamide groups in chitin using a strong NaOH solution at high temperatures (100°C or higher) that results in chitosan production. Chemical deacetylation is widely used since it is affordable and allows for mass manufacturing, but it also has drawbacks, such as higher energy consumption and higher air pollutants owing to alkaline conditions. An alternate enzymatic technique employing chitin deacetylases has been investigated to circumvent these constraints in the manufacture of chitosan. The use of chitin deacetylases to convert chitin to chitosan allows for the synthesis of innovative, well-defined chitosan in a controlled, non-degradable manner. This approach is very useful for producing chitosan oligomers [87,146]. Using bacterial, fungal, and viral chitin deacetylases, they created various potential partly acetylated chitosan tetramers with a certain degree of acetylation and acetylation pattern. The availability of these completely developed tetramers expands the scope of research into chitosan structure-function relationships [164].

Due to their huge molecular weight, low solubility, and high viscosity of chitosan solutions, chitin and chitosan have restricted biological uses. Chito-oligosaccharide, which is formed when chitin and chitosan are broken down, is thought to be a superior substitute [142]. Chitin deacetylase (EC 3.5.1.41) catalyzes the

hydrolysis of N-acetamido bonds in chitin to produce chitosan [165]. The *Mucor rouxii* fungal extract's deacetylase action was the first active enzyme discovered and partially purified in the mid-1970s. Many more chitin deacetylases and chito-oligosaccharide deacetylases (EC 3.5.1.105) were discovered and isolated from a diverse range of organisms, including archaea, marine bacteria, fungi, and insects [166,167]. The enzymes are either excreted in the periplasmic zone or in the culture media, exhibiting outstanding thermal stability at an optimal temperature of 50°C and show a very high specificity for the N-acetyl-d-glucosamine polymers associated with β -(1, 4). The performance of enzymes is pH-dependent. The pH of enzymes affects their performance. It is critical to highlight that the chitin deacetylases generated by *Colletotrichum lindemuthianum* and *Aspergillus nidulans* are not inhibited by acetate and are hence suitable for future biotechnological applications. The comparison of chemical and biological methods in chitosan preparation from seafood waste is summarized as depicted in Fig. 6.

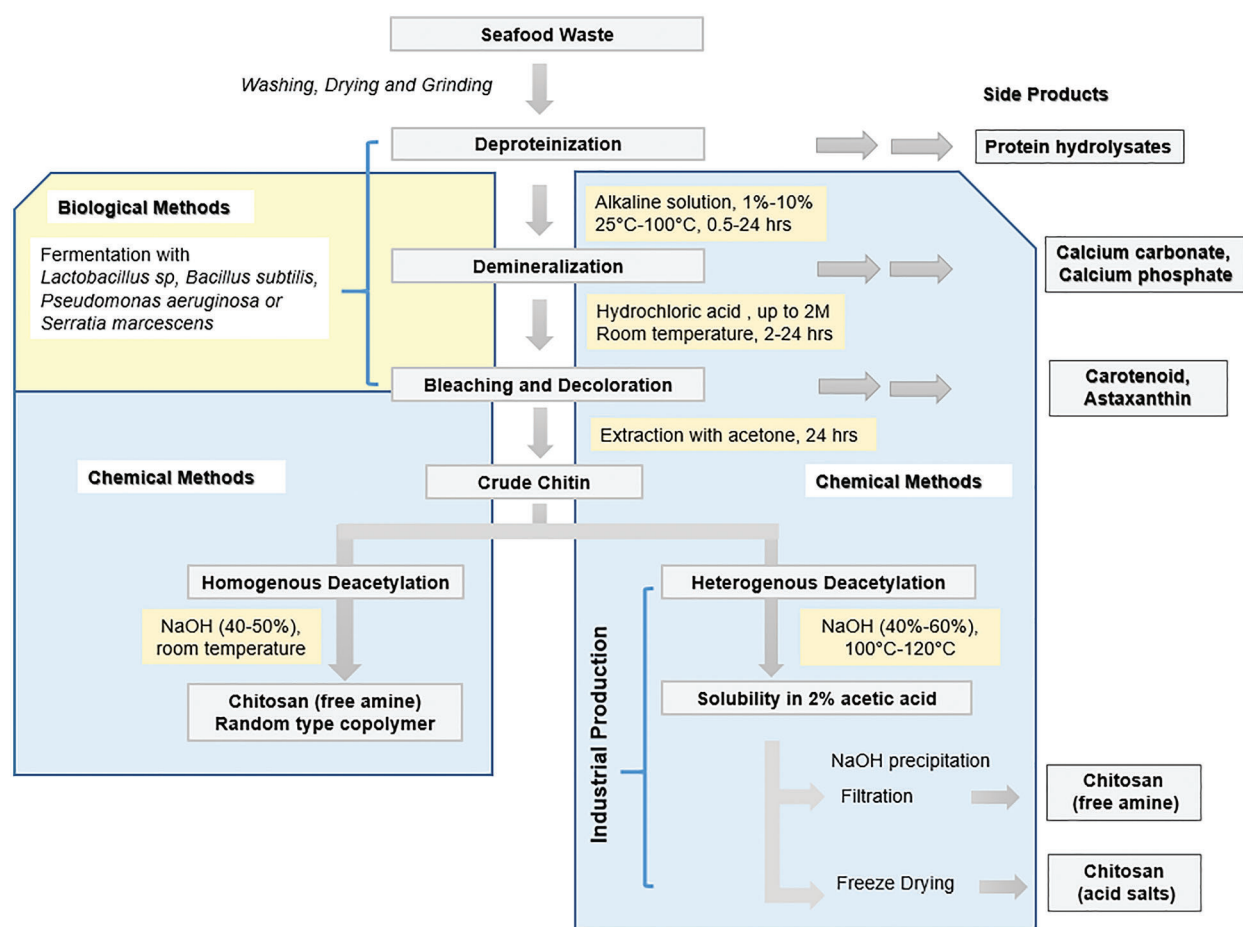


Figure 6: Comparison biological and chemical methodology for chitosan production

5.2 Bioeconomy of Chitin and Chitosan

Chitosan research started a long time ago. The PubMed database recorded the first publication related to chitosan in the year 1953. Since then, the number of research publications exploring the potential of chitosan in different applications, and ways of increasing its production has increased, as shown in Fig. 7. One of the key elements fueling the growing interest in chitosan research is the abundant supply of raw materials. The global demand for chitosan-based biomaterials is currently dominated by North America and Europe, where

the products are increasingly used for biomedical and agricultural industries. Asia Pacific is fast turning into another major user and manufacturer of chitosan-based biomaterials, boosting the demand for chitosan-based biomaterials in the region.

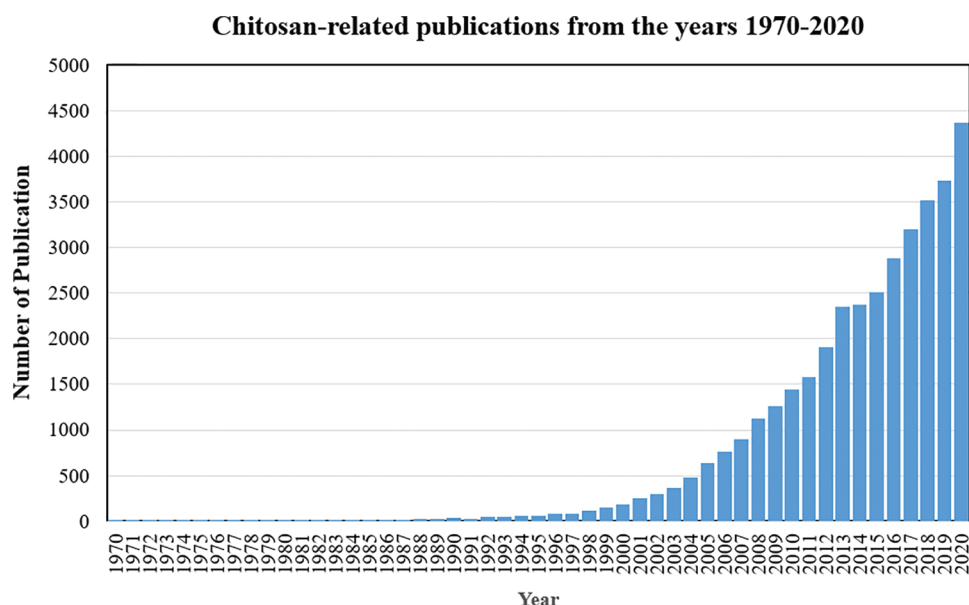


Figure 7: Chitosan-related publications from the years 1970 to 2020. The first paper on chitosan was published in the year 1953, and since then, growing research interest in chitosan has resulted in an exponential increase in the number of publications

The global demand for chitosan derivatives is expected to grow at a CAGR (compound annual growth rate) of around 6.3% over the next 5 years. As of the year 2015, the global chitosan market size was reported at USD 1,205 million. This amount has increased by more than 300%, reaching USD 5.71 billion in 2018, and is expected to grow at a CAGR of 20.8% through 2025. The expected growth is mainly due to the increasing application of chitosan in various industries, such as agriculture, water treatment industry, medical, dental, cosmetic, pharmaceutical, and food [138,168,169].

Applications of chitosan are in diverse areas. Chitosan is ideal for agricultural application with its biodegradable and non-toxic characteristics. It is essential for improving germination, leaf formation, seed harvesting, soil moisture retention, and increasing crop production. The demand for chitosan-based biomaterials is also driven by the growth of the water treatment industry. As the population of countries such as China, India, and Bangladesh in the Asia-Pacific region grows, the demand for clean drinking water has increased. Countries like India and China are facing the problem of water pollution. It is estimated that this situation will increase the demand for chitosan-based biomaterials in the region by improving the consumption of water treatment compounds. The promising potential of chitosan in the field of biological applications is encouraging research activities. Chitosan-based biomaterials are used in a wide range of medical industry applications such as orthopedics and periodontal applications, cancer detection, regenerative medicine, and medication delivery, as depicted in Fig. 5 [8,56–62,170–172].

The unique features of chitosan, such as biocompatibility, biodegradability, non-toxic properties, bio-adhesion, and muco-adhesiveness, are important in the medical field. The need for chitosan-based biomedical applications is expected to grow as new technologies emerge. It is expected that ongoing

research activities will focus on discovering new applications of chitosan and developing existing technologies, which will further contribute to the growth of the chitosan market [94,96].

6 Current Research Trends on Chitosan

Chitosan, an economical multifunctional polymer with several desirable physicochemical and biological characteristics, obtainable from seafood waste, has been investigated for use in agriculture, pharmacy, and biomedicine. In the past two years, research has focused on exploring the potential of chitosan application, especially within the areas of the food industry [173–175], health care [176–178], and environmental application [179,180]. Through its numerous functional groups, chitosan allows for diverse forms of interaction, such as ionic hydrogen bonding, which is one of the reasons for its appeal in a variety of applications [181,182]. This study presented tremendous potential for researchers to develop an efficient and environmentally friendly methodology for chitosan extraction and production in innovative ways.

The increasing demand of consumers for food quality and safety has led to increasing research activities in the areas of food production and packaging. Amongst the many factors contributing to food quality and safety, microorganism contamination has drawn much attention. In the past, chemical food preservatives such as sodium nitrite, potassium sorbate, and sodium benzoate were used in the food industry to prevent microbial growth [183]. Over time, the need to explore alternatives to these preservatives has grown due to the increasing concerns about the safety of these chemical preservatives for long-term consumption. The alternative will need to at least fulfill the criteria of cost effectiveness and food safety, in addition to high antimicrobial activity. Chitosan, as well as chitosan film coating, enables it to increase the shelf life of food. Therefore, it is the most recent targeted material for this purpose [173–175,184]. Edible film with a visible colorimetric indicator, which changes color when the pH value changes due to rotting, is a promising direction for food quality management [185]. Another contemporary development in extending food shelf life is the development of biodegradable polymer packaging materials. Polylactic acid with chitosan nanoparticles that are used as plasticizers with polyvinyl alcohol and polyethylene glycol as the link agents has a substantial improvement in tensile strength and antimicrobial action against aerobic microorganisms. An amphiphilic nanoparticle made from a polylactic acid filler with chitosan grafting is a more flexible material and greatly improves the characteristics of the polylactic acid fillers [186].

Chitosan is one of the popular materials for the development of therapeutic and diagnostic applications and also in the area of nanomedicine [176–178] to improve drug delivery efficiency and drug therapeutic index [187,188]. Nanoparticle products for the diagnosis of cancer imaging as well as for pharmacotherapeutic treatment have been established, and the fabrication of nanoparticle systems by applying chitosan has been explored. Today, the use of chitosan in medicinal treatments as nanocarriers and chitosan-based nanoformulations is still under investigation [189,190].

Chitosan is also playing a very important role in the area of sensor development, especially in the topic of biological evaluation. The versatility of chitosan in forming various forms of structures such as films, microgel/hydrogel and nanocomposites, and its adhesive properties are the main reasons for its being a popular candidate for sensor development [191,192]. These chitosan nanocomposite-based biosensors have shown high sensitivity, selectivity, and stability for detecting a variety of targets, ranging from biomolecules, DNA, and microorganisms. It expands possibilities in the field of biosensor research [193].

The combination of the biological and chemical properties of chitosan allows for a wide range of applications. The versatility of structural formations has enabled application not only for water treatment but also for air and soil treatments [179,180,182]. Research has proven that chitosan-based adsorbents work well for the removal of organic and inorganic pollutants. Therefore, chitosan modification by various physical and chemical processes has emerged as one of the promising strategies for the removal of contaminants from aqueous medium [194,195].

The potential of chitosan has never been exhausted, and it still has a lot of room to grow. All of this was made possible by the enormous possibilities provided by the combination of its chemical and biological characteristics. The possibility of chitosan applications, therefore, will only be limited by the creativity of the research.

7 Conclusion

Southeast Asia, with its high consumption of fish and seafood per capita, produces tons of fishery and crustacean waste. These fishery and crustacean waste share common chemical compounds such as calcium carbonate, chitin, protein, and lipids. Therefore, seafood waste extractable compounds carry huge economic potential, generally in the area of agriculture and pharmaceuticals. The economic value of these compounds has driven the increasing trend in the number of research on these compounds, especially chitin and its derivative, chitosan. The global market for chitosan increased more than 300% from the year of 2015 to 2018 and is projected to continue at a CARG value of 20.8% until 2025.

Research activities on chitin and chitosan encompass different areas. Some researchers focus on process improvement to meet current needs. In recent years, ensuring environmental safety in preparation and purification of chitin and chitosan has become one of the hot factors to be considered in research. Cost-effective and time saving efforts have always been the factors to be considered in designing the methodology by the researchers. Exploring the potential uses of chitin and chitosan is another popular area of research. Potential innovative applications of chitin and chitosan in the field of biomedical and pharmaceutical remains as one of the popular areas of research. To date, the potential of chitin and chitosan remains wide open. Interest in exploring the potential of these compounds for diverse applications can be reflected in the numbers of publications and filed/granted patents, as pointed out in this article. Researchers that continuously contributed their efforts to chitin and chitosan research often shared common goals. These efforts, in turn, help turn seafood waste into wealth by generating novel materials for human wellbeing.

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